# **& Catalysis**

## Highly Enantioselective Hydrogenation of  $\beta$ -Ketoenamides with the Rh-ZhangPhos Catalyst

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#### **S** Supporting Information

[AB](#page-2-0)STRACT: [A series of](#page-2-0)  $\beta$ -ketoenamides have been successfully synthesized and applied in the enantioselective hydrogenation with the Rh-ZhangPhos catalyst. This methodology provides an efficient access to a variety of optically active  $\beta$ -amino ketones with up to 99% enantiomeric excess (ee).  $\beta$ -Amino ketones can be converted to chiral γ-arylamines in one step without loss of enantioselectivity. Furthermore, direct hydrogenation of  $\beta$ -ketoenamide to chiral 1,3amino alcohol is also tested with extremely high enantioselectivity and diastereoselectivity.



KEYWORDS: asymmetric catalysis, enantioselectivity, hydrogenation, phosphine ligand, β-ketoenamides

The synthesis of  $\beta$ -amino ketones has drawn great attention<br>because of its importance in biomedical research<sup>1−5</sup> and<br>the pharmacoutical industry.<sup>6–10</sup>  $\beta$  Amino, latence are also the pharmaceutical industry.<sup>6−10</sup>  $\beta$ -Amino ketones can also serve as key precursors for the synthesis of amino alcoh[ols,](#page-2-0)<sup>11-14</sup> 1,3-diamines,<sup>15−17</sup> and  $\beta$ -am[ino](#page-2-0) acids. Several stoichiometric and catalytic methods have already been reported fo[r](#page-2-0) t[he](#page-2-0) synthesis of  $\beta$ -a[m](#page-2-0)ino ketones.<sup>18−22</sup> Notable examples include Lewis acid mediated hetero-Michael addition reactions<sup>23-27</sup> and Mannich-type reactions.<sup>28-31</sup> [A](#page-2-0)symmetric synthesis of  $\beta$ amino ketones from sulfinimines was also reported by [Da](#page-2-0)v[is](#page-2-0)' group.<sup>32,33</sup> We have reported [enan](#page-2-0)tioselective hydrogenation of  $\beta$ -ketoenamides as an efficient way to prepare enantiomerically pure  $\beta$ [-am](#page-2-0)ino ketones using the Rh-DuanPhos system.<sup>34,35</sup> Very recently, we reported the design and synthesis of a new electron rich bisphosphine ligand, Zhangphos, which coul[d be](#page-2-0) synthesized in five steps from commercially available trans-1,2cyclohexanedicarboxylic acid (Scheme  $1$ ).<sup>36</sup> Both enantiomers of ZhangPhos are available with asymmetric synthesis. Chiral cyclohexane structure on the backbone [is](#page-2-0) also believed to further increase the electron-donating ability and conformational rigidity of the ligand. Herein, we report a highly enantioselective hydrogenation of  $\beta$ -ketoenamides with the Rh-ZhangPhos catalyst.  $β$ -ketoenamides were prepared in one step





on multigram scale via direct condensation of readily accessible 1,3-diketones with acetamide (Scheme 2). Only (Z)-enamide  $(4)$  was observed in all cases.<sup>37,38,35</sup>

#### Scheme 2. Preparation of  $β$ [-Ketoen](#page-2-0)amides



Our initial study began with the hydrogenation of 4a as the model substrate in MeOH under ambient hydrogen pressure with 0.5 mol % Rh-ZhangPhos complex (Table 1). To our delight, the reaction can be finished within 30 min with complete conversion into chiral  $\beta$ -amino ketone [5a](#page-1-0) and with high enantioselectivity (96% enantiomeric excess (ee)). Solvent screening revealed that ethanol and isopropanol are also reliable solvents under the same reaction conditions. Aprotic solvents such as toluene and tetrahydrofuran (THF) caused dramatic loss of the reactivity (entry 5 and entry 8). Further increasing hydrogen pressure caused a slight drop in enantioselectivity (entry 11). Lowering the catalyst loading to 0.2 mol % resulted in lower conversion (entry 12).

To further explore the efficiency of the Rh-ZhangPhos catalytic system, we attempted the asymmetric hydrogenation of a series of substrates (4b−4m) under the optimized conditions in MeOH (Table 2). All substrates were hydrogenated in full conversions with excellent enantioseletivities. For example, substrates bearin[g b](#page-1-0)oth para-substituted electron-

Received: February 16, 2012 Revised: April 25, 2012 Published: May 15, 2012



<span id="page-1-0"></span>Table 1. Rhodium-Catalyzed Asymmetric Hydrogenation of 4a under Various Conditions

	NHAc	$[Rh(NBD)$ (S)-ZhangPhos]BF <sub>4</sub> $\Omega$	<b>NHAc</b>	
Ph	Me	$H_2$ , solvent	Ьµ.	Me
4а				5a
entry <sup>a</sup>	$P_{\rm H2}$ [atm]	solvent	conv. $\lbrack\% \rbrack^b$	ee $\lceil \% \rceil^c$
$\mathbf{1}$	1.0	MeOH	>99	96
$\overline{2}$	1.0	EtOH	>99	94
3	1.0	<sup>i</sup> PrOH	>99	95
4	1.0	EtOAc	86	95
5	1.0	toluene	<5	n.d.
6	1.0	CH <sub>2</sub> Cl <sub>2</sub>	64	96
7	1.0	ClCH <sub>2</sub> CH <sub>2</sub> Cl	13	91
8	1.0	<b>THF</b>	<5	n.d.
9	1.0	1,4-dioxane	35	93
10	2.0	MeOH	>99	96
11	5.0	MeOH	>99	95
$12^d$	1.0	MeOH	68	95

a Unless otherwise noted, all reactions were carried out with a substrate/catalyst ratio of 200:1 at room temperature for 30 min. because, called the car contract temperature for so main product. <sup>c</sup>The ee value of 5a was determined by chiral GC analysis. The absolute configuration of 5a was assigned by comparison of the observed optical rotation with reported data. <sup>d</sup> The reaction was carried out with a substrate/catalyst ratio of 500:1 at room temperature for 30 min.

Table 2. Asymmetric Hydrogenation of 4 with the Rh-ZhangPhos Catalyst

	NHAc	[Rh(NBD)(S)-ZhangPhos]BF <sub>4</sub>	<b>NHAc</b>	
R١	$R^2$	R <sup>1</sup> 1 atm $H_2$ , MeOH		5
$entry^a$	substrate	R <sup>1</sup>	$R^2$	ee(%) <sup>b</sup> (config.) <sup>c</sup>
$\mathbf{1}$	4a	$C_6H_5$	Me	96(S)
$\mathfrak{p}$	4b	$p$ -Me $C_6H_4$	Me	$96(-)$
3	4c	$p$ -FC <sub>6</sub> H <sub>4</sub>	Me	$98(-)$
4	4d	$p$ -ClC <sub>6</sub> H <sub>4</sub>	Me	$98 (+)$
5	4e	$p$ -Br $C_6H_4$	Me	$97 (+)$
6	4f	$p$ -MeOC <sub>6</sub> H <sub>4</sub>	Me	$98 (+)$
7	4g	$p$ - <sup>t</sup> BuC <sub>6</sub> H <sub>4</sub>	Me	$96(-)$
8	4h	$p$ -CyC <sub>6</sub> H <sub>4</sub>	Me	$95(-)$
9	4i	$m$ -Me $C_6H_4$	Me	$98(-)$
10	4j	2-thienyl	Me	$99 (+)$
11	4k	2-naphthyl	Me	$93(-)$
12	41	Me	Me	$99(-)$
13 <sup>d</sup>	4m	$C_6H_5$	Et	$90(-)$

a Unless otherwise noted, all reactions were carried out with a substrate/catalyst ratio of 200:1 in MeOH at room temperature under 1 atm of  $H_2$  for 30 min. In all cases, 100% conversion was observed. not expect be mind in an easely recive centrement was esserted.<br>Determined by chiral GC or HPLC analysis. "The absolute configurations of  $5b-5m$  were not determined.  $d_t = 24 h$ 

donating and withdrawing groups on the aromatic ring were hydrogenated with uniformly high enantioselectivities (95− 98%, entries 2−8). Meta-substituted moiety on the aromatic ring further increased the enantioselectivity (entry 9). Substrates with thienyl and naphthyl groups also afforded the hydrogenation products with up to 99% ee (entry 10 and 11). A simple aliphatic substrate like 4l could be hydrogenated under the same reaction conditions with 99% ee. However, increasing the steric bulkiness of the  $R^2$  group resulted in significant erosion of the reactivity and enantioselectivity (entry 13). Chiral 1,3-amino alcohol was observed as the hydrogenation product when increasing the hydrogen pressure to 20 atm and reaction time to 24 h. Excellent enantioselectivity and diastereoselectivity were recorded with 4a as substrate in EtOAc (Scheme 3).





Chiral  $\beta$ -amino ketones such as 5a could simply be converted to γ-arylamine by Pd/C-catalyzed hydrogenation without loss of enantioselectivity (Scheme 4).39−<sup>43</sup> This transformation provided an reliable catalytic approach to chiral  $\gamma$ -arylamines, which is highly pharmaceutically an[d](#page-2-0) [bio](#page-2-0)logically valuable.<sup>44-48</sup>





In conclusion, we have developed an efficient enantioselective hydrogenation of a wide range of  $\beta$ -ketoenamides using the Rh-ZhangPhos catalyst system. This method provided an efficient access to a variety of optically active  $\beta$ -amino ketones with excellent enantioselectivities. Further reduction of  $\beta$ -amino ketones could give a variety of protected chiral γ-aryl amines.

#### **EXPERIMENTAL SECTION**

General Procedure for the Substrate Preparation. A toluene solution (150 mL) of substituted 1,3-diketone (50 mmol), acetamide (250 mmol), and a catalytic amount of  $p$ -TsOH (10 mmol) was charged in a Dean−Stark apparatus and refluxed for 24 h. After the solution was cooled down to room temperature, the solvent was evaporated, and the concentrated mixture was passed through a flash chromatography column filled with silica gel (eluent: EtOAc/hexane). The product 4 was collected as a stable crystalline solid.

Preparation of Rh-ZhangPhos Complex [Rh(NBD)- **ZhangPhos]BF<sub>4</sub>.** To a solution of  $[Rh(NBD)_2]BF_4$  (67.3 mg, 0.18 mmol) in degassed THF (1 mL) at −10 °C was added a solution of (1S,1′S,2R,2′R,3aS,3′aS,7aS,7′aS)-2 (ZhangPhos) (74.6 mg, 0.189 mmol) in THF (2 mL). The resulting red solution was allowed to warm to room temperature and stirred for 15 min. The solution was concentrated to about 1 mL and then was added degassed  $Et<sub>2</sub>O (12 mL)$  under vigorous stirring. The resulting precipitate was filtered, further washed with ether  $(3 \times 10 \text{ mL})$  for three times, and dried under vacuum to afford a brown solid (79.1 mg, 65%).

General Procedure for the Asymmetric Hydrogenation. In a glovebox filled with nitrogen, [Rh(NBD)- ZhangPhos] $BF_4$  complex (6.7 mg, 0.01 mmol) was dissolved in corresponding solvent (10 mL). To 1 mL of this solution, the substrate (0.2 mmol for 0.5 mol % catalyst loading) was

<span id="page-2-0"></span>added. The resulting solution was then transferred into an autoclave and charged with hydrogen. The hydrogenation was performed at room temperature for 30 min. After carefully releasing the pressure in hood, the reaction mixture was passed through a short silica-gel plug to remove the catalyst. Enantiomeric excess were determined with the resulting solution by chiral GC or HPLC.

#### ■ ASSOCIATED CONTENT

#### **S** Supporting Information

Further details on the NMR characterization and enantioselectivity analysis of the products. This material is available free of charge via the Internet at http://pubs.acs.org.

■ AUTHOR INFORMATI[ON](http://pubs.acs.org)

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#### Funding

This wor[k was supported by th](mailto:xumu@rci.rutgers.edu)e National Institutes of Health (GM58832).

#### Notes

The authors declare no competing financial interest.

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